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E1-26 The room volume is $(21\ 13\ 12)\text{ft}^3(0:3048\ \text{m}/\text{ft})^3 = 92:8\text{m}^3$. The mass contained in the room is $(92:8\text{m}^3)(1:21\text{kg}=\text{m}^3) = 112\text{kg}$: E1-27 One mole of sugar cubes would have a volume of $N_A\ 1:0\ \text{cm}^3$, where N_A is the Avogadro constant. Since the volume of a cube is equal to the length cubed, $V = l^3$, then $l = \sqrt[3]{N_A\ \text{cm}^3} = 8:4\ 10^7\ \text{cm}$. E1-28 The number of seconds in a week is

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